

A METHOD TO INTRODUCE A 3,3-DIMETHYLALLYL UNIT ORTHO TO A PHENOL

R.D.H. Murray, M.M. Ballantyne and K.P. Mathai

Department of Chemistry, University of Glasgow, Glasgow W.2.

(Received in UK 4 December 1969; accepted for publication 19 December 1969)

The 3,3-dimethylallyl grouping, present as such or as an oxygenated form, is a structural feature commonly found in coumarins¹ and other compounds of natural origin^{1,2}; hence methods for synthesis of this residue are of importance. One method³, developed by Späth⁴ for the synthesis of osthenol (4) involves the reaction of a substituted salicylaldehyde with 3,3-dimethylallyl bromide followed by Perkin condensation to the coumarin. Direct ortho C-dimethylallylation of 7-hydroxycoumarins has only been successful⁵ (~50%) when a 6-OMe group is also present and normally it results^{6,7} in low yields (~10%) of either the desired product or the corresponding dimethylchromanocoumarin.

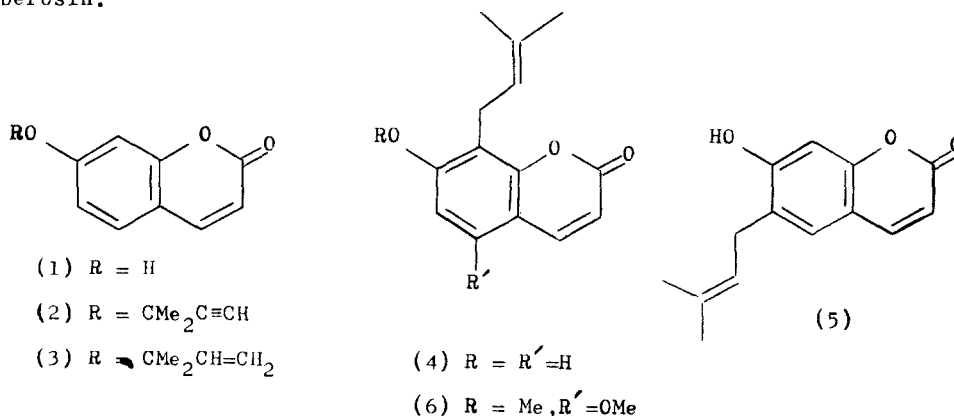
When we found⁸ that Claisen rearrangement of a 3,3-dimethylallyl ether afforded the ortho-(1,1-dimethylallyl)hydroxycoumarin, we surmised that ortho-(3,3-dimethylallyl)hydroxycoumarins could be synthesised in an analogous manner from the corresponding 1,1-dimethylallyl ethers. A compound of this type, 7-O-(1,1-dimethylallyl)scopoletin, had already been synthesised⁹ by reduction of the 1,1-dimethylpropargyl ether. Recently the preparation of such ethers has been reported¹⁰ together with a method² for reductive ring opening to the ortho-(3,3-dimethylallyl)phenol from the derived chromenes¹⁰.

The synthesis of osthenol (4) was attempted via the above Claisen rearrangement. Umbelliferone (1) on treatment with 3-chloro-3-methylbut-1-yne, K₂CO₃ and KI in refluxing aqueous acetone was converted to the 1,1-dimethylpropargyl ether¹¹ (2, m.p. 136-139°, 77%), which on hydrogenation afforded (3) (m.p. 76-78°, 96%). Pyrolysis of (3) at 130° gave both osthenol⁴ (4, m.p. 129-131°, 74%) and demethylsuberosin^{7,12} (5, m.p. 133-134°, 14%). In a similar manner, coumurrayin^{3,6} (6, m.p. 155.5-157°) has been prepared with

a comparable overall yield.

This sequence seems to be a useful method for the introduction of an isoprenyl residue ortho to a phenolic hydroxyl group and its scope is being investigated.

We are most grateful to Professor R.A. Raphael for suggesting the route to the allylic ethers and to Dr. T.J. King for a generous gift of demethyl-suberosin.



1. F.M. Dean, Naturally Occurring Oxygen Ring Compounds, Butterworths (1963).
2. A.J. Birch, M. Maung and A. Pelter, Aust. J. Chem., **22**, 1923 (1969).
3. P.W. Austin, T.R. Seshadri, M.S. Sood and Vishwapaul, Tetrahedron, **24**, 3247 (1968).
4. E. Späth and H. Holzen, Ber., **67**, 264 (1934).
5. F.A.L. Anet, G.K. Hughes and E. Ritchie, Aust. J. Sci. Res., **A2**, 608 (1969); S.F. Dykes, W.D. Ollis, M. Sainsbury and J.S.P. Schwarz, Tetrahedron, **20**, 1331 (1964).
6. D.L. Dreyer, J. Org. Chem., **33**, 3574 (1968); H. Tanino and S. Inoue, Chem. Pharm. Bull., **17**, 1071 (1969).
7. P.W. Austin and T.R. Seshadri, Indian J. Chem., **6**, 412 (1968).
8. M.M. Ballantyne, R.D.H. Murray and A.B. Penrose, Tetrahedron Letters, 4155 (1968).
9. K.A.M. Gillies, B.Sc. thesis, Glasgow, 1967; R.D.H. Murray and K.A.M. Gillies, unpublished results.
10. J. Hlubucek, E. Ritchie and W.C. Taylor, Tetrahedron Letters, 1369 (1969).
11. All coumarins gave satisfactory analytical and spectral data.
12. F.E. King, J.R. Housely and T.J. King, J. Chem. Soc., 1392 (1954).